# **Supporting Information:**

# Theory for Gas Solubility and Hydrophobic Interaction in Aqueous Electrolyte Solutions

Ryuichi Okamoto $^{a*}$  and Kenichiro Koga $^a$ 

<sup>a</sup>Research Institute for Interdisciplinary Science, Okayama University, Okayama 700-8530, Japan

\* E-mail: okamoto-ryuichi@okayama-u.ac.jp

# SI-A Inversion of $I^{ij}$ for small $n_{\rm h}$

For small  $n_{\rm h}$  we can easily invert the  $3 \times 3$  matrix  $I^{ij}$  to obtain  $I_{ij}$ . First, from Eq. (7), we obtain  $I^{ij} = I^{ij}_{\rm s} + O(n_{\rm h})$  for  $i, j = {\rm w, I}$ ,  $I^{i{\rm h}} = \partial \nu_{\rm h}^{\rm s}/\partial n_i + O(n_{\rm h})$  for  $i = {\rm w, I}$ , and  $I^{\rm hh} = 1/n_{\rm h} + U^{\rm s}_{\rm hh} + O(n_{\rm h})$ . Here  $I^{ij}_{\rm s}$  has been defined below Eq. (27). We expand  $I_{ij}$  as

$$I_{ij} = I_{ij}^{s} + O(n_h) \quad (i, j = w, I)$$
 (SI-1)

$$I_{ih} = A_i n_h + O(n_h^2) \quad (i = w, I)$$
 (SI-2)

where  $2 \times 2$  matrix  $\{I_{ij}^s\}_{i,j=\text{w,I}}$  is the inverse matrix of  $\{I_{\text{s}}^{ij}\}_{i,j=\text{w,I}}$ . In Eq. (SI-2) we can easily confirm the zeroth order term identically vanishes. The relation  $\sum_j I_{\text{w}j} I^{j\text{h}} = \sum_j I_{\text{I}j} I^{j\text{h}} = 0$  then yields

$$A_i = -\sum_{j=w,I} I_{ij}^s (\partial \nu_h^s / \partial n_j).$$
 (SI-3)

Inverting this, we obtain

$$\partial \nu_{\rm h}^{\rm s}/\partial n_i = -\sum_{j={\rm w},{\rm I}} I_{\rm s}^{ij} A_j$$
 (SI-4)

$$= \begin{cases} \Delta_0^{-1}(-I_{\text{II}}^{\text{s}}A_{\text{w}} + I_{\text{wI}}^{\text{s}}A_{\text{I}}) & (i = \text{w}) \\ \Delta_0^{-1}(I_{\text{wI}}^{\text{s}}A_{\text{w}} - I_{\text{ww}}^{\text{s}}A_{\text{I}}) & (i = \text{I}). \end{cases}$$
(SI-5)

Substituting Eq. (SI-5) into Eq.(34) and using Eqs. (15) and (29), we obtain

$$g_{\rm h} = (1/n^2 \chi_{\rm I})(A_{\rm w} n_{\rm I} - A_{\rm I} n_{\rm w}).$$
 (SI-6)

Similarly, substitution of Eq. (SI-4) into (21) yields

$$v_{\rm h}^{\rm s*} = -(v_{\rm w}^{\rm s} A_{\rm w} + v_{\rm I}^{\rm s} A_{\rm I}),$$
 (SI-7)

where use has been made of Eq. (15). We can readily invert Eqs. (SI-6) and (SI-7) to have

$$A_{\mathbf{w}} = n^2 \chi_{\mathbf{I}} g_{\mathbf{h}} v_{\mathbf{I}}^{\mathbf{s}} - n_{\mathbf{w}} v_{\mathbf{h}}^{\mathbf{s}*} \tag{SI-8}$$

$$A_{\rm I} = -n^2 \chi_{\rm I} g_{\rm h} v_{\rm w}^{\rm s} - n_{\rm I} v_{\rm h}^{\rm s*}. \tag{SI-9}$$

Next, neglecting the terms of order  $n_h$  in  $\sum_i I^{hi} I_{ih} = 1$ , we obtain

$$1/I_{\rm hh} = 1/n_{\rm h} + U_{\rm hh}^{\rm s}/k_{\rm B}T + \sum_{i=\rm w,I} A_i (\partial \nu_{\rm h}^{\rm s}/\partial n_i)$$
$$= 1/n_{\rm h} + U_{\rm hh}^{\rm s}/k_{\rm B}T - \sum_{i=\rm w,I} I_{\rm s}^{ij} A_i A_j, \tag{SI-10}$$

where Eq. (SI-4) has been used in the second line. Setting  $n_h = 0$  in Eqs. (15) and (17), we have

$$\sum_{i,j=w,I} I_{ij}^{s} v_{i}^{s} v_{j}^{s} = k_{B} T \kappa_{T}^{s} = 1 / \sum_{i,j=w,I} I_{s}^{ij} n_{i} n_{j}.$$
 (SI-11)

Substituting Eqs. (SI-8) and (SI-9) into (SI-10) and using (15) and (SI-11), we obtain

$$1/I_{\rm hh} = 1/n_{\rm h} + U_{\rm hh}^{\rm eff}/k_{\rm B}T,$$
 (SI-12)

where the terms of order  $n_{\rm h}$  have been neglected.

# SI-B Calculation of $C_{\rm I}^{(1)}$

In this section we can set  $n_{\rm h}=0$ . From Eq. (25) we find

$$(\partial n_{\mathbf{w}}/\partial X_{\mathbf{I}})_{T,p} = -n^2 v_{\mathbf{I}}^{\mathbf{s}}, \quad (\partial n_{\mathbf{I}}/\partial X_{\mathbf{I}})_{T,p} = n^2 v_{\mathbf{w}}^{\mathbf{s}}.$$
 (SI-13)

Then for an arbitrary function  $w(n_{\rm w}, n_{\rm I})$ , we have

$$\lim_{n_{\rm I}\to 0} \left(\frac{\partial w}{\partial X_{\rm I}}\right)_{T,p} = n_{\rm w} \frac{\partial w}{\partial n_{\rm I}} \Big|_{n_{\rm I}\to 0} - n_{\rm w}^2 v_{\rm I} \frac{\partial w}{\partial n_{\rm w}} \Big|_{n_{\rm I}\to 0}.$$
 (SI-14)

From Eq. (10), we have  $(\kappa_T^s)^{-1} = k_B T \sum_{i,j=w,I} n_i n_j I_s^{ij}$ . Substituting  $w = (\kappa_T^s)^{-1}$  into Eq. (SI-14), we obtain

$$\lim_{n_{\mathrm{I}}\to 0} \left(\frac{\partial (\kappa_{T}^{\mathrm{s}})^{-1}}{\partial X_{\mathrm{I}}}\right)_{T,p} = k_{\mathrm{B}} T n_{\mathrm{w}} (1 + 2n_{\mathrm{w}} I_{0}^{\mathrm{wI}} + n_{\mathrm{w}}^{2} I_{0}^{\mathrm{wwI}}) - k_{\mathrm{B}} T n_{\mathrm{w}}^{2} v_{\mathrm{I}} [\partial (n_{\mathrm{w}}^{2} I_{0}^{\mathrm{ww}}) / \partial n_{\mathrm{w}}]$$

$$= n_{\mathrm{w}} \kappa_{\mathrm{w}}^{-1} [\partial (v_{\mathrm{I}} n_{\mathrm{w}}) / \partial n_{\mathrm{w}}], \qquad (\mathrm{SI-15})$$

where we have defined  $I_0^{\text{w}i} \equiv \lim_{n_{\text{I}} \to 0} I_{\text{s}}^{\text{w}i}$  and  $I_0^{\text{w}ij} \equiv \lim_{n_{\text{I}}, n_{\text{h}} \to 0} (\partial I^{\text{w}i}/\partial n_j) = I_0^{\text{w}ji}$ . This readily yields

$$\lim_{n_{\rm I} \to 0} (\partial \kappa_T^{\rm s} / \partial X_{\rm I})_{T,p} = -n_{\rm w} \kappa_{\rm w} [\partial (v_{\rm I} n_{\rm w}) / \partial n_{\rm w}]. \tag{SI-16}$$

Similarly, from Eqs. (15) and (SI-14), we obtain

$$\lim_{n_{\rm I}\to 0} \left(\frac{\partial v_{\rm h}^{\rm s}}{\partial X_{\rm I}}\right)_{T,p} = (v_{\rm h}/\kappa_{\rm w}) \lim_{n_{\rm I}\to 0} \left(\frac{\partial \kappa_{T}^{\rm s}}{\partial X_{\rm I}}\right)_{T,p} + k_{\rm B}T\kappa_{\rm w} \left[n_{\rm w}(I_{0}^{\rm hI} + n_{\rm w}I_{0}^{\rm hwI}) - n_{\rm w}^{2}v_{\rm I}(I_{0}^{\rm hw} + n_{\rm w}I_{0}^{\rm hww})\right]$$

$$= k_{\rm B}T\kappa_{\rm w}n_{\rm w} \frac{\partial \left[n_{\rm w}(K_{\rm S}/2 - v_{\rm I})\right]}{\partial n_{\rm w}}.$$
(SI-17)

Using Eqs. (18) and (SI-16), we have

$$\lim_{n_{\rm I} \to 0} \left( \frac{\partial v_{\rm h}^{\rm s*}}{\partial X_{\rm I}} \right)_{T,p} = k_{\rm B} T \kappa_{\rm w} n_{\rm w} \frac{\partial (n_{\rm w} K_{\rm S}/2)}{\partial n_{\rm w}}. \tag{SI-18}$$

Substitution of  $w=U_{\rm hh}^{\rm s}$  into Eq. (SI-14) yields

$$\lim_{n_{\rm I}\to 0} \left(\frac{\partial U_{\rm hh}^{\rm s}}{\partial X_{\rm I}}\right)_{T,p} = k_{\rm B} T (n_{\rm w} I_0^{\rm hhI} - n_{\rm w}^2 v_{\rm I} I_0^{\rm hhw}), \tag{SI-19}$$

where we have defined  $I_0^{\text{hh}i} \equiv \lim_{n_{\text{I}},n_{\text{h}}\to 0} (\partial I^{\text{hh}}/\partial n_i) = \lim_{n_{\text{I}},n_{\text{h}}\to 0} (\partial I^{\text{h}i}/\partial n_{\text{h}})$  for i=w,I.Using Eqs. (SI-15), (SI-17), and (SI-19), one can calculate  $C_{\text{I}}^{(1)}$  in Eq. (41).

# SI-C Expressions for salts of general valence numbers

For a salt with general valence numbers, i.e.,  $X_a Y_b \to a X^{Z_1} + b Y^{Z_2}$  with  $b = -a Z_1/Z_2$ , the relation between the salt density  $n_{\rm s}$  and the ion density  $n_{\rm I}$  in Eq. (5) becomes  $n_{\rm s} = n_{\rm I}/\mathcal{N}$  with  $\mathcal{N} = a + b = a(1 - Z_1/Z_2)$ . In this case Eq. (40) becomes

$$C_{\rm I} \equiv -\lim_{n_{\rm I} \to 0} \left(\frac{\partial B}{\partial n_{\rm S}}\right)_{T,p} = \frac{-\mathcal{N}/2}{k_{\rm B}Tn_{\rm w}} \lim_{n_{\rm I} \to 0} \left(\frac{\partial U_{\rm hh}^{\rm eff}}{\partial X_{\rm I}}\right)_{T,p}.$$
 (SI-20)

Furthermore, Eqs. (31) and (35) are generalized to

$$K_{\rm S} = (\mathcal{N}/k_{\rm B}T)(U_{\rm hI} - v_{\rm h}^* v_{\rm I}/\kappa_{\rm w}) \tag{SI-21}$$

$$\lim_{n_1 \to 0} g_{\rm h} = n_{\rm w} K_{\rm S} / \mathcal{N}, \tag{SI-22}$$

while Eq. (28) remains unchanged. Then Eq. (42) becomes

$$C_{\rm I}^{(2)} \equiv \frac{\mathcal{N}}{2n_{\rm w}} \lim_{n_{\rm I} \to 0} (\partial [\chi_{\rm I} g_{\rm h}^2] / \partial X_{\rm I})_{T,p} = K_{\rm S}^2 / (2\mathcal{N}).$$
 (SI-23)

### SI-D Long-wavelength density fluctuations

We discuss the long-wavelength density fluctuations, which shall make the physical meanings of  $\chi_{\rm I}$  and  $U_{\rm hh}^{\rm eff}$  (or B) clearer. The Bjerrum length  $\ell_B$  and the Debye wavenumber  $\kappa$  are defined as

$$\ell_B = e^2/k_{\rm B}T\varepsilon, \quad \kappa = \sqrt{4\pi\ell_B n_{\rm I}}.$$
 (SI-24)

As in the previous paper, S1 we study equilibrium density fluctuations on the scales larger than the Debye length  $\kappa^{-1}$ .

The deviations of the densities from their average values are denoted by  $\delta n_i$ ,

$$\delta \hat{n}_i(\mathbf{r}) \equiv \hat{n}_i(\mathbf{r}) - \langle \hat{n}_i \rangle \quad (i = w, c, a, h).$$
 (SI-25)

where  $\langle \hat{n}_i \rangle$ 's are given by Eqs. (4) and (5). We expand the free energy functional  $\mathcal{F}$  defined in Eq. (43) with respect to  $\delta \hat{n}_i$  about the homogeneous equilibrium values  $\langle \hat{n}_i \rangle$ . The second order term, which governs the density fluctuations, is given by

$$\delta^{2} \mathcal{F} = \frac{1}{2} \int \sum_{i,j=\text{w.c.a.h}} \hat{f}_{ij} \delta \hat{n}_{i} \delta \hat{n}_{j} d\mathbf{r} + \int \frac{\varepsilon |\nabla \Phi|^{2}}{8\pi} d\mathbf{r}, \qquad (SI-26)$$

where  $\hat{f}_{ij} = \partial^2 \hat{f}/\partial \hat{n}_i \partial \hat{n}_j|_{\{\hat{n}_k = \langle \hat{n}_k \rangle\}}$ . The probability distribution of the density fluctuations is proportional to  $\exp[-\delta^2 \hat{F}/k_{\rm B}T]$ .

For any fluctuating variables  $\delta \hat{A}(\mathbf{r})$  and  $\delta \hat{B}(\mathbf{r})$  of zero mean  $\langle \delta \hat{A} \rangle = \langle \delta \hat{B} \rangle = 0$ , we define

$$\langle \delta \hat{A} \circ \delta \hat{B} \rangle \equiv \int d\mathbf{r} \langle \delta \hat{A}(\mathbf{r}) \delta \hat{B}(0) \rangle.$$
 (SI-27)

Let  $\delta \hat{A}_{\boldsymbol{q}}$  and  $\delta \hat{B}_{\boldsymbol{q}}$  be the Fourier components of  $\delta \hat{A}$  and  $\delta \hat{B}$ , respectively, where  $\boldsymbol{q}$  is the wave vector. Then we have  $^{\mathrm{S2}} \lim_{\boldsymbol{q} \to 0} \langle \delta \hat{A}_{\boldsymbol{q}} \delta \hat{B}_{-\boldsymbol{q}} \rangle / V = \langle \delta \hat{A} \circ \delta \hat{B} \rangle$ .

#### SI-D.1 Fluctuations in electrolyte solvent without solute

We first study the density fluctuations in the electrolyte solvent without the hydrophobic solute,

$$n_{\rm h}=0.$$

Then  $\hat{f}$  in Eq. (43) is replaced by  $\hat{f}^{s}(\hat{n}_{w}, \hat{n}_{c}, \hat{n}_{a}) \equiv \hat{f}(\hat{n}_{w}, \hat{n}_{c}, \hat{n}_{a}, 0)$ . In the previous paper, S1 the ion-specific fluctuations in dilute electrolyte solutions have been studied. Here we slightly generalize the previous theory to a one that is (at least formally) valid for more concentrated electrolyte solutions.

It will turn out that the following combinations are convenient:

$$\delta\hat{\phi} \equiv \sum_{i=\text{w.c.a}} v_i^{\text{s}} \delta \hat{n}_i, \quad \delta \hat{X}_{\rho} \equiv n^{-1} (\delta \hat{n}_{\text{c}} - \delta \hat{n}_{\text{a}}),$$
 (SI-28)

$$\delta \hat{X}_{\rm I} \equiv n^{-2} [n_{\rm w} (\delta \hat{n}_{\rm c} + \delta \hat{n}_{\rm a}) - n_{\rm I} \delta \hat{n}_{\rm w}]. \tag{SI-29}$$

Here,  $\delta \hat{\phi}$  is the volume deviation,  $\delta \hat{X}_{\rho}$  the deviation of the ion molar fraction difference, and  $\delta \hat{X}_{\rm I}$  the deviation of the total ion molar fraction. For an electrolyte solvent without solute Eqs. (44) and (SI-26), respectively, become

$$f^{\rm s}(n_{\rm w}, n_{\rm I}) = \hat{f}^{\rm s}(n_{\rm w}, n_{\rm I}/2, n_{\rm I}/2)$$
 (SI-30)

$$\delta^{2} \mathcal{F} = \frac{1}{2} \int \sum_{i,j=\text{w,c,a}} \hat{f}_{ij}^{\text{s}} \delta \hat{n}_{i} \delta \hat{n}_{j} d\mathbf{r} + \int \frac{\varepsilon |\nabla \Phi|^{2}}{8\pi} d\mathbf{r}.$$
 (SI-31)

where  $\hat{f}_{ij}^{s} = \partial^{2}\hat{f}/\partial\hat{n}_{i}\partial\hat{n}_{j}|_{\{\hat{n}_{k}=\langle n_{k}\rangle\}}$ . Inverting Eqs. (SI-28) and (SI-29), we can express  $\{\delta\hat{n}_{i}\}_{i=w,c,a}$  as linear combinations of  $\hat{\phi}$ ,  $\delta\hat{X}_{\rho}$ , and  $\delta\hat{X}_{I}$ . Substituting the result into  $\sum_{i,j=w,c,a}\hat{f}_{ij}^{s}\delta\hat{n}_{i}\delta\hat{n}_{j}$  and using Eqs. (45)–(47), we find

$$\sum_{i,j} \hat{f}_{ij}^{s} \delta \hat{n}_{i} \delta \hat{n}_{j} = \delta \hat{\phi}^{2} / \kappa_{T}^{s} + \sum_{i,j=\rho,I} \gamma_{ij} \delta \hat{X}_{i} \delta \hat{X}_{j},$$
 (SI-32)

where the terms proportional to  $\delta \hat{\phi} \delta \hat{X}_i$  ( $i = \rho, I$ ) identically vanish. We do not discuss the coefficients  $\gamma_{\rho\rho}$  and  $\gamma_{\rho I}$  as they are not relevant to the fluctuations in the long wavelength

limit. The coefficient  $\gamma_{\rm II}$  is calculated as

$$\gamma_{\text{II}} = n^{4} \Big[ (v_{\text{I}}^{\text{s}})^{2} \hat{f}_{\text{ww}}^{\text{s}} + (v_{\text{w}}^{\text{s}}/2)^{2} \sum_{i,j=\text{c,a}} \hat{f}_{ij}^{\text{s}} - v_{\text{I}}^{\text{s}} v_{\text{w}}^{\text{s}} \sum_{i=\text{c,a}} \hat{f}_{\text{w}i}^{\text{s}} \Big] 
= k_{\text{B}} T n^{4} \Big[ (v_{\text{I}}^{\text{s}})^{2} I_{\text{s}}^{\text{ww}} + (v_{\text{w}}^{\text{s}})^{2} I_{\text{s}}^{\text{II}} - 2 v_{\text{I}}^{\text{s}} v_{\text{w}}^{\text{s}} I_{\text{s}}^{\text{Iw}} \Big] 
= k_{\text{B}} T n^{4} \Delta_{0}^{-1} \sum_{i,j=\text{w,I}} v_{i}^{\text{s}} v_{j}^{\text{s}} I_{ij}^{\text{s}} 
= k_{\text{B}} T \chi_{\text{I}}^{-1}.$$
(SI-33)

Here the second line follows from Eq. (SI-30), and the last line from Eqs. (SI-11) and (29). The second term of Eq. (SI-31) is rewritten as

$$\int \frac{\varepsilon |\nabla \Phi|^2}{8\pi} d\mathbf{r} = k_{\rm B} T \int \frac{d\mathbf{q}}{(2\pi)^3} 2\pi \ell_B n^2 q^{-2} |\delta \hat{X}_{\rho \mathbf{q}}|^2$$
 (SI-34)

The long-range Coulombic interaction is bilinear only in  $\delta \hat{X}_{\rho}$ , and thus from Eq. (SI-32) we obtain

$$\langle \delta \hat{X}_{\rho \mathbf{q}} \delta \hat{\phi}_{-\mathbf{q}} \rangle = \langle \delta \hat{X}_{\mathbf{I} \mathbf{q}} \delta \hat{\phi}_{-\mathbf{q}} \rangle = 0 \tag{SI-35}$$

$$\langle |\delta \hat{\phi}_{\mathbf{q}}|^2 \rangle = k_{\rm B} T V \kappa_T^{\rm s}.$$
 (SI-36)

Furthermore, as  $\mathbf{q} \to 0$ , the coefficient in front of  $|\delta \hat{X}_{\rho \mathbf{q}}|^2$  diverges, suppressing the fluctuations of  $\delta \hat{X}_{\rho \mathbf{q}}$ . Therefore, combining Eqs. (SI-32)–(SI-34), we obtain

$$\langle \delta \hat{X}_{\rm I} \circ \delta \hat{X}_{\rm I} \rangle = \chi_{\rm I} \tag{SI-37}$$

$$\langle \delta \hat{X}_{\rho} \circ \delta \hat{X}_{\rho} \rangle = \langle \delta \hat{X}_{\rho} \circ \delta \hat{X}_{I} \rangle = \langle \delta \hat{X}_{\rho} \circ \delta \hat{\phi} \rangle = 0. \tag{SI-38}$$

where  $\chi_{\rm I}$  is the solvent composition susceptibility defined below Eq. (26). The fluctuation variance in Eq. (SI-37) has the same form as the composition fluctuation variance in a non-ionic binary solvent. S3 From Eq. (SI-38), we have  $\langle \delta \hat{X}_{\rho} \circ \delta \hat{A} \rangle = 0$ , where  $\delta \hat{A}$  is an arbitrary

linear combination of  $\delta\hat{\phi}$ ,  $\delta\hat{X}_{\rho}$ , and  $\delta\hat{X}_{\rm I}$  (hence, of  $\delta\hat{n}_{\rm w}$ ,  $\delta\hat{n}_{\rm c}$ , and  $\delta\hat{n}_{\rm a}$ ). In particular, setting  $\delta\hat{A} = \delta\hat{n}_i$  ( $i = {\rm w, c, a}$ ), we obtain the well-know relations

$$\langle \delta \hat{n}_{c} \circ \delta \hat{n}_{c} \rangle = \langle \delta \hat{n}_{a} \circ \delta \hat{n}_{a} \rangle = \langle \delta \hat{n}_{c} \circ \delta \hat{n}_{a} \rangle$$
 (SI-39)

$$\langle \delta \hat{n}_{\mathbf{w}} \circ \delta \hat{n}_{\mathbf{c}} \rangle = \langle \delta \hat{n}_{\mathbf{w}} \circ \delta \hat{n}_{\mathbf{a}} \rangle. \tag{SI-40}$$

The result in Eqs. (SI-36), (SI-37), and (SI-38) indicates that in the limit  $\mathbf{q} \to 0$  the probability distribution of  $\delta \hat{\phi}_{\mathbf{q}}$  and  $\delta \hat{X}_{\mathbf{I}\mathbf{q}}$  is proportional to  $\exp[-\delta^2 \hat{f}_{\mathbf{q}}^{\mathbf{s}}/k_{\mathrm{B}}T]$  with

$$\delta^2 \hat{f}_{\mathbf{q}}^{\mathrm{s}} = (1/2V) \left[ |\delta \hat{\phi}_{\mathbf{q}}|^2 / \kappa_T^{\mathrm{s}} + k_{\mathrm{B}} T |\delta \hat{X}_{\mathrm{I}\mathbf{q}}|^2 / \chi_{\mathrm{I}} \right]. \tag{SI-41}$$

Note that the above expression can be obtained by minimizing  $\delta^2 \hat{F}$  in Eq. (SI-31) with respect to  $\delta \hat{X}_{\rho q}$  for  $q \to 0$ .

For small ion densities, we may expand the local free energy for the electrolyte solvent  $\hat{f}^{\rm s}(\hat{n}_{\rm w},\hat{n}_{\rm c},\hat{n}_{\rm a})$  as S1

$$\hat{f}^{s} = f_{w}(\hat{n}_{w}) + k_{B}T \sum_{i=c,a} \hat{n}_{i} \{ \ln(\hat{n}_{i}\lambda_{i}^{3}) - 1 + \nu_{i}(\hat{n}_{w}) \}$$

$$- \frac{k_{B}T}{12\pi} \hat{\kappa}^{3} + \frac{1}{2} \sum_{i=c,a} U_{ij}(\hat{n}_{w}) \hat{n}_{i} \hat{n}_{j} + \cdots .$$
(SI-42)

In the above, the third term ( $\propto \hat{\kappa}^3$ ) is the DH free energy that arises from the ion density fluctuations on the scales smaller than the Debye length, where  $\hat{\kappa}$  (without solute) is defined in accordance with Eq. (SI-24) as

$$\hat{\kappa}(\hat{n}_{w}, \hat{n}_{c}, \hat{n}_{a}) = [4\pi \ell_{B}(\hat{n}_{w})(\hat{n}_{c} + \hat{n}_{a})]^{1/2}.$$
 (SI-43)

The effective ion-ion (short-range) interaction, which is the counterpart to Eqs. (1) and (32),

is defined as

$$U_{\rm II}^{\rm eff} \equiv \frac{1}{4} \sum_{i,j={\rm c.a}} \left( U_{ij} - \frac{v_i^* v_j^*}{\kappa_{\rm w}} \right) = U_{\rm II} - \frac{(v_{\rm I}^*)^2}{\kappa_{\rm w}},$$
 (SI-44)

where  $U_{\rm II} = \sum_{i,j=\rm c,a} U_{ij}/4$ . As shown in the previous paper, S1  $U_{\rm II}^{\rm eff}$  is the key quantity to explain the ion-specificity in the salt activity coefficient, osmotic coefficient, and the deviation coefficient for the salt partial volume<sup>1</sup>.

From Eqs. (46) and (SI-42), the ionic partial volumes in the dilution limit  $n_{\rm I} \to 0$  are given by

$$v_i^* = k_{\rm B} T \kappa_{\rm w} n_{\rm w} \nu_i' \quad (i = c, a, I), \tag{SI-45}$$

where  $\nu_{\rm I} = (\nu_{\rm c} + \nu_{\rm a})/2$  and  $\nu_i'(n_{\rm w}) = \partial \nu_i(n_{\rm w})/\partial n_{\rm w}$ . For small  $n_{\rm I}$ , Eqs. (SI-42) and (SI-30) yield

$$I_{\rm s}^{\rm ww} = 1/k_{\rm B}T n_{\rm w}^2 \kappa_{\rm w} + O(n_{\rm I}) \tag{SI-46}$$

$$I_{\rm s}^{\rm II} = 1/n_{\rm I} + U_{\rm II}/k_{\rm B}T - \kappa^3/16\pi n_{\rm I}^2 + O(n_{\rm I})$$
 (SI-47)

$$I_{\rm s}^{\rm wI} = \nu_{\rm I}' + O(n_{\rm I}^{1/2}),$$
 (SI-48)

where  $U_{\rm I}$  and  $\nu_{\rm I}$  are defined below Eqs. (SI-44) and (SI-45), respectively. We also have  $v_{\rm I}^{\rm s} = v_{\rm I} + O(n_{\rm I}^{1/2})$  and  $v_{\rm w}^{\rm s} = (1 - v_{\rm I}n_{\rm I})/n_{\rm w} + O(n_{\rm I}^{3/2})$ , where the latter follows from Eq. (17). Substituting these and Eqs. (SI-46)–(SI-48) into Eq. (27), we obtain

$$\chi_{\rm I}^{-1} \simeq \frac{n^3}{n_{\rm w} n_{\rm I}} \left[ \frac{1}{2\chi} - (v_{\rm I} + v_{\rm I}^*) n_{\rm I} + \frac{n_{\rm I}}{n_{\rm w}} \right]$$
(SI-49)

with  $\chi^{-1} = 2 - \ell_B \kappa / 2 + 2 n_{\rm I} U_{\rm II}^{\rm eff} / k_{\rm B} T$ . In the square bracket of Eq. (SI-49), the terms of order  $n_{\rm I}^{3/2}$  have been neglected. Substitution of Eqs. (SI-28), (SI-29), and (SI-49) into Eq. (SI-41)

To be precise, the defined in the previous paper S1 were  $U_{\text{eff}}$  and U, that were twice the present definitions, i.e.,  $U_{\text{eff}} = 2U_{\text{II}}^{\text{eff}}$  and  $U = 2U_{\text{II}}$ .

yields

$$\delta^2 \hat{f}_{\mathbf{q}}^{\mathrm{s}} \simeq (1/2V) \left[ |\delta \hat{\phi}_{v\mathbf{q}}|^2 / \kappa_{\mathrm{w}} + k_{\mathrm{B}} T |\delta \hat{n}_{\mathrm{I}\mathbf{q}}|^2 / 2n_{\mathrm{I}} \chi \right], \tag{SI-50}$$

where  $\delta \phi_v = n_{\rm w}^{-1} \delta \hat{n}_{\rm w} + v_{\rm c}^* \delta \hat{n}_{\rm c} + + v_{\rm a}^* \delta \hat{n}_{\rm a}$  and  $\delta \hat{n}_{\rm I} = \delta \hat{n}_{\rm c} + \delta \hat{n}_{\rm a}$ . From Eqs.(SI-39) and (SI-50), we reproduce the previous result for dilute electrolyte solutions, S1

$$\langle \delta \hat{n}_{c} \circ \delta \hat{n}_{c} \rangle = \langle \delta \hat{n}_{a} \circ \delta \hat{n}_{a} \rangle = \langle \delta \hat{n}_{c} \circ \delta \hat{n}_{a} \rangle \simeq n_{I} \chi / 2.$$
 (SI-51)

#### SI-D.2 Solute-solute correlation

Now we add a small amount of hydrophobic solute. In accordance with Eqs. (7) and (44), we have

$$\hat{f} = \hat{f}^{s} + k_{B}T\hat{n}_{h}\{\ln(\hat{n}_{h}\lambda_{h}^{3}) - 1 + \hat{\nu}_{h}^{s}(\hat{n}_{w}, \hat{n}_{c}, \hat{n}_{a})\} 
+ \frac{1}{2}\hat{U}_{hh}^{s}(\hat{n}_{w}, \hat{n}_{c}, \hat{n}_{a})\hat{n}_{h}^{2} + \cdots,$$
(SI-52)

where  $\hat{\nu}_{\rm h}^{\rm s}$  and  $\hat{U}_{\rm hh}^{\rm s}$  satisfies

$$\hat{\nu}_{\rm h}^{\rm s}(n_{\rm w}, n_{\rm I}/2, n_{\rm I}/2) = \nu_{\rm h}^{\rm s}(n_{\rm w}, n_{\rm I})$$
 (SI-53)

$$\hat{U}_{\rm bb}^{\rm s}(n_{\rm w}, n_{\rm I}/2, n_{\rm I}/2) = U_{\rm bb}^{\rm s}(n_{\rm w}, n_{\rm I}). \tag{SI-54}$$

We define the total volume deviation  $\delta \hat{\phi}_{\rm tot}$  as  $^{\rm S3}$ 

$$\delta\hat{\phi}_{\text{tot}} \equiv \sum_{i=\text{w.c.a}} v_i^s \delta\hat{n}_i + v_h^{\text{s*}} \delta\hat{n}_h = \delta\hat{\phi} + v_h^{\text{s*}} \delta\hat{n}_h.$$
 (SI-55)

Expanding Eq. (SI-52) with respect to the density deviations  $\delta \hat{n}_i$ , and neglecting the

terms of order  $n_{\rm h}$ , we obtain

$$\sum_{i,j=\text{w,c,a,h}} \hat{f}_{ij}\delta\hat{n}_i\delta\hat{n}_j = \sum_{i,j=\text{c,a}} \hat{f}_{ij}^{\text{s}}\delta\hat{n}_i\delta\hat{n}_j + \delta^2\hat{f}^{\text{h}}.$$
 (SI-56)

Here the second term is the bilinear form including  $\delta \hat{n}_h$ ,

$$\delta^2 \hat{f}^{\mathrm{h}} = (k_{\mathrm{B}} T / n_{\mathrm{h}} + U_{\mathrm{hh}}^{\mathrm{s}}) \delta \hat{n}_{\mathrm{h}}^2 + 2k_{\mathrm{B}} T \sum_{i=\mathrm{w.c.a}} \hat{\nu}_{\mathrm{h}i}^{\mathrm{s}} \delta \hat{n}_i \delta \hat{n}_{\mathrm{h}}$$
 (SI-57)

with  $\hat{\nu}_{hi}^{s} = \partial \hat{\nu}_{h}^{s} / \partial \hat{n}_{i}|_{\{\hat{n}_{i} = \langle \hat{n}_{i} \rangle \}}$ . Using Eqs. (21), (34), (SI-28), (SI-29) (SI-32), (SI-33) (SI-56), and (SI-57), we find

$$\sum_{i,j=\text{w,c,a,h}} \hat{f}_{ij} \delta \hat{n}_{i} \delta \hat{n}_{j} = \frac{\delta \hat{\phi}_{\text{tot}}^{2}}{\kappa_{T}^{\text{s}}} + \frac{k_{\text{B}} T (\delta \hat{X}_{\text{I}} + \chi_{\text{I}} g_{\text{h}} \delta \hat{n}_{\text{h}})^{2}}{\chi_{\text{I}}} + (k_{\text{B}} T / n_{\text{h}} + U_{\text{hh}}^{\text{eff}}) \delta \hat{n}_{\text{h}}^{2} + \gamma_{\rho\rho} \delta \hat{X}_{\rho}^{2} + 2\gamma_{\rho\text{I}} \delta \hat{X}_{\rho} \delta \hat{X}_{\text{I}} + \gamma_{\rho\text{h}} \delta \hat{X}_{\rho} \delta \hat{n}_{\text{h}}$$
(SI-58)

where we have introduced the coefficient  $\gamma_{\rho h}$ , but it is irrelevant to the long wavelength fluctuations. As in SI-D.1, the long-range electrostatic interaction suppresses the fluctuations of  $X_{\rho}$  for small  $\boldsymbol{q}$ . Hence, in the limit  $\boldsymbol{q} \to 0$ , Eqs. (SI-38), (SI-39) and (SI-40) still hold in the presence of solute. In addition we have  $\langle \delta \hat{X}_{\rho} \circ \delta \hat{n}_{h} \rangle = 0$ , or, equivalently,

$$\langle \delta \hat{n}_{\rm h} \circ \delta \hat{n}_{\rm c} \rangle = \langle \delta \hat{n}_{\rm h} \circ \delta \hat{n}_{\rm a} \rangle.$$
 (SI-59)

The long-wavelength  $(\boldsymbol{q} \to 0)$  fluctuations of  $\delta \hat{\phi}_{\rm tot}$ ,  $\delta \hat{X}_{\rm I}$ , and  $\delta \hat{n}_{\rm h}$  are governed by the probability distribution proportional to  $\exp[-\delta^2 \hat{f}_{\boldsymbol{q}}/k_{\rm B}T]$  with

$$\delta^{2} \hat{f}_{\boldsymbol{q}} = (1/2V) \left[ \frac{|\delta \hat{\phi}_{\text{tot}\boldsymbol{q}}|^{2}}{\kappa_{T}^{\text{s}}} + \frac{k_{\text{B}}T|\delta \hat{X}_{\text{I}\boldsymbol{q}} + \chi_{\text{I}}g_{\text{h}}\delta \hat{n}_{\text{h}\boldsymbol{q}}|^{2}}{\chi_{\text{I}}} + |\delta \hat{n}_{\text{h}\boldsymbol{q}}|^{2} (k_{\text{B}}T/n_{\text{h}} + U_{\text{hh}}^{\text{eff}}) \right], \quad (\text{SI-60})$$

where the terms of order  $n_{\rm h}$  have been neglected, and  $U_{\rm hh}^{\rm eff}$  has been defined in Eq. (39). The

fluctuation variance of the solute is then given by

$$\langle \delta \hat{n}_{\rm h} \circ \delta \hat{n}_{\rm h} \rangle = (1/n_{\rm h} + U_{\rm bh}^{\rm eff}/k_{\rm B}T)^{-1}.$$
 (SI-61)

This result also follows from Eq. (SI-12) and the well-known relation  $I_{\rm hh} = \langle \delta \hat{n}_{\rm h} \circ \delta \hat{n}_{\rm h} \rangle$  (see also sentences below Eq. (39)). The results in Eqs. (39), (SI-60) and (SI-61) are the same as the previous ones for a ternary mixture composed of a non-ionic binary solvent and a solute; S3 this is because in the present ionic case the long wavelength fluctuations of  $\delta \hat{X}_{\rho}$  are suppressed by the long-range Coulombic interaction, and thus in the limit  $q \to 0$  the cations and anions can be treated as a single indistinguishable component. S1,S4

# SI-E Kirkwood-Buff integrals

Let  $g_{ij}(r)$  denote the radial distribution function associated with species i and j. Its space integral is called the Kirkwood-Buff integral (KBI):  $^{S5}G_{ij} \equiv 4\pi \int_0^\infty dr \, r^2 [g_{ij}(r) - 1]$ . Since in Eqs. (SI-39), (SI-40) and (SI-59) cations and anions are indistinguishable, the cations and anions are regarded as indistinguishable, single species when KBIs are discussed in the literature.  $^{S6-S13}$  In the present case we define the KBIs for species w (water), I (ion), and h (solute). For convenience we define  $\delta \hat{n}_1 \equiv \delta \hat{n}_c + \delta \hat{n}_a$ . The well-known relation  $I_{ij} = \langle \delta \hat{n}_i \circ \delta \hat{n}_j \rangle$  combines the thermodynamics and the KBIs. Integrating the general relation  $\langle \delta \hat{n}_i(\mathbf{r})\delta \hat{n}_j(0)\rangle = n_i n_j [g_{ij}(r) - 1] + n_i \delta_{ij}\delta(\mathbf{r})$ , we have

$$I_{ij} = \langle \delta \hat{n}_i \circ \delta \hat{n}_j \rangle = n_i n_j G_{ij} + n_i \delta_{ij}$$
 (SI-62)

First we discuss KBIs for the electrolyte solvent without solute. From Eqs. (SI-37)–(SI-40) and (SI-62) (or, from Eqs. (29) and (SI-62)), we obtain

$$\chi_{\rm I} = n_{\rm w} n_{\rm I} / n^3 + (n_{\rm w}^2 n_{\rm I}^2 / n^4) (G_{\rm ww}^{\rm s} + G_{\rm II}^{\rm s} - 2G_{\rm wI}^{\rm s}).$$
 (SI-63)

where  $G_{ij}^s$ 's are KBIs evaluated at  $n_h = 0$ . The determinant  $\Delta_0 = \det\{I_{ij}^s\}$  can be expressed in terms of KBIs. Hence, The isothermal compressibility  $\kappa_T^s$  can also be expressed in terms of KBIs by using Eqs. (29) and (SI-63). At  $n_h = 0$ , Eqs. (15), (29) and (SI-62) yield

$$v_{\rm I}^{\rm s} = (n_{\rm I} n_{\rm w}^2 / n^4 \chi_{\rm I})(n_{\rm w}^{-1} + G_{\rm ww}^{\rm s} - G_{\rm wI}^{\rm s})$$
 (SI-64)

$$v_{\rm w}^{\rm s} = (n_{\rm I}^2 n_{\rm w}/n^4 \chi_{\rm I})(n_{\rm I}^{-1} + G_{\rm II}^{\rm s} - G_{\rm wI}^{\rm s}).$$
 (SI-65)

For a vanishing salt density  $n_{\rm I} \to 0$  we substitute Eq. (SI-49) into (SI-64), and obtain

$$v_{\rm I} = -G_{\rm wI}^0 + k_{\rm B}T\kappa_{\rm w}, \quad v_{\rm I}^* = -G_{\rm wI}^0,$$
 (SI-66)

where  $G_{ij}^0$  is the KBI in the limit  $n_{\rm h}, n_{\rm I} \to 0$ . In the same limit, Eq. (SI-65) correctly reduces to Eq. (23). These formal results for electrolyte solvents are exactly the same as those for non-ionic binary fluids, <sup>S3</sup> but in the present ionic case one needs to be careful when taking the limit  $n_{\rm I} \to 0$  for ion-ion KBI,  $G_{\rm II}^{\rm s}$ . For small  $n_{\rm I}$  Eq. (SI-51) yields <sup>S1,S14</sup>  $G_{\rm II}^{\rm s} \simeq (2\chi - 1)/n_{\rm I} \simeq \ell_B \kappa/4n_{\rm I} \propto n_{\rm I}^{-1/2}$ , which diverges as  $n_{\rm I} \to 0$ ; this divergence stems from the Debye-Hückel free energy, which is specific to electrolyte solutions.

Now we discuss the KBIs in the presence of a small amount of solute. First, Eqs. (SI-61) and (SI-62) yield

$$G_{\rm hh} = -\frac{U_{\rm hh}^{\rm eff}/k_{\rm B}T}{1 + n_{\rm h}U_{\rm hh}^{\rm eff}/k_{\rm B}T} = \frac{G_{\rm hh}^{\rm s}}{1 - n_{\rm h}G_{\rm hh}^{\rm s}},\tag{SI-67}$$

where  $G_{\rm hh}^{\rm s} = -U_{\rm hh}^{\rm eff}/k_{\rm B}T$  is the solute-solute KBI in the limit  $n_{\rm h} \to 0$ . The relation (SI-67) shows that KBI  $G_{\rm hh}$  at finite  $n_{\rm h}$  noticeably deviates from its dilute limit  $G_{\rm hh}^{\rm s}$  when  $|G_{\rm hh}^{\rm s}| \sim n_{\rm h}^{-1}$ ; this observation is sometimes not negligible when one attempts to compute  $G_{\rm hh}^{\rm s}$  from MD simulations of small but finite  $n_{\rm h}$ . Using Eq. (38), we obtain

$$U_{\rm hh}^{\rm eff}/k_{\rm B}T = -G_{\rm hh}^{\rm s} = 2B.$$
 (SI-68)

Next, Eqs. (SI-62) and (SI-2) yield  $A_i = G_{ih}^s n_i$  (i = w, I). Substituting this into Eqs. (SI-6) and (SI-7), we obtain

$$g_{\rm h} = (n_{\rm w} n_{\rm I} / n^2 \chi_{\rm I}) (G_{\rm wh}^{\rm s} - G_{\rm Ih}^{\rm s})$$
 (SI-69)

$$v_{\rm h}^{\rm s*} = -(v_{\rm w}^{\rm s} n_{\rm w} G_{\rm wh}^{\rm s} + v_{\rm I}^{\rm s} n_{\rm I} G_{\rm Ih}^{\rm s}).$$
 (SI-70)

In the limit  $n_{\rm I} \to 0$ , these reduce to

$$K_{\rm S} = 2(G_{\rm wh}^0 - G_{\rm Ih}^0), \quad v_{\rm h}^* = -G_{\rm wh}^0$$
 (SI-71)

where use has been made of Eqs. (35) and (SI-49) in deriving the first relation. Finally, Eqs. (33) and (SI-71) yield

$$U_{\rm hI}^{\rm eff}/k_{\rm B}T = -G_{\rm Ih}^0. \tag{SI-72}$$

# SI-F Debye-McAulay theory and Eq. (52)

Debye and McAulay<sup>S15</sup> have studied effective solute-salt (electrostatic) interaction on the basis of Born theory. S16 The idea is that the effective solute-salt interaction  $U_{\rm hs}^{\rm DM}$  is given by the Born energy difference  $\Delta\mu_{\rm Born}=U_{\rm hs}^{\rm DM}n_{\rm h}+O(n_{\rm h}^2)$  between a pair of ions in pure water and that in water containing the solute. One can expand the dielectric permittivity with respect to  $n_{\rm h}$ ,  $\varepsilon=\varepsilon^0(1+Dn_{\rm h}+O(n_{\rm h}^2))$ , where  $\varepsilon^0$  is the dielectric permittivity of pure water and D is a coefficient of proportionality. Then from the Born model one readily obtain

$$U_{\rm hs}^{\rm DM} = -(k_{\rm B}T\ell_BD/2)(R_{\rm c}^{-1} + R_{\rm a}^{-1}).$$
 (SI-73)

In isobaric condition and without salt, meanwhile, the water density varies according to  $(\partial n_{\rm w}/\partial n_{\rm h})_{T,p} = -\bar{v}_{\rm h}/\bar{v}_{\rm w}$  as the solute density varies (we can derive this relation as in the

same manner for Eq. (25)). We thus have  $n_{\rm w}-n_{\rm w}^0=-v_{\rm h}n_{\rm h}n_{\rm w}^0+O(n_{\rm h}^2)$  where  $n_{\rm w}^0$  is the water density without solute. Since we have assumed the dielectric permittivity explicitly depends only on  $n_{\rm w}$ , we expand  $\varepsilon(n_{\rm w})=\varepsilon^0+(\partial\varepsilon/\partial n_{\rm w})(n_{\rm w}-n_{\rm w}^0)+\cdots$ . We thus have

$$D = -a_{\varepsilon}v_{\rm h},\tag{SI-74}$$

where  $a_{\varepsilon} = n_{\rm w} \varepsilon^{-1} (\partial \varepsilon / \partial n_{\rm w})$ . Substituting Eq. (SI-74) into (SI-73) and using Eq. (53), we obtain

$$U_{\rm hs}^{\rm DM} = -2v_{\rm h}v_{\rm I}^{\rm B}/\kappa_{\rm w} \simeq -2v_{\rm h}^*v_{\rm I}^{\rm B}/\kappa_{\rm w}. \tag{SI-75}$$

The right hand side is indeed the same as  $K_{\rm S}^{\rm el}$  in Eq. (52) with  $v_{\rm I}^{\rm el}$  being replaced by  $v_{\rm I}^{\rm B}$ .

#### References

- (S1) Okamoto, R.; Koga, K.; Onuki, A. Theory of electrolytes including steric, attractive, and hydration interactions. *J. Chem. Phys.* **2020**, *153*, 074503.
- (S2) Onuki, A. Phase Transition Dynamics; Cambridge, 2002.
- (S3) Okamoto, R.; Onuki, A. Theory of nonionic hydrophobic solutes in mixture solvent: Solvent-mediated interaction and solute-induced phase separation. J. Chem. Phys. 2018, 149, 014501.
- (S4) Smith, P. E. Cosolvent Interactions with Biomolecules: Relating Computer Simulation Data to Experimental Thermodynamic Data. J. Phys. Chem. B 2004, 108, 18716– 18724.
- (S5) Kirkwood, J. G.; Buff, F. P. The statistical mechanical theory of solutions. I. J. Chem. Phys. 1951, 19, 774–777.

- (S6) Ben-Naim, A. Molecular theory of solutions; Oxford, 2006.
- (S7) Smith, P. E.; Mazo, R. M. On the Theory of Solute Solubility in Mixed Solvents. *J. Phys. Chem. B* **2008**, *112*, 7875–7884.
- (S8) Katsuto, H.; Okamoto, R.; Sumi, T.; Koga, K. Ion Size Dependences of the Salting-Out Effect: Reversed Order of Sodium and Lithium Ions. J. Phys. Chem. B 2021, 125, 6296–6305.
- (S9) Shimizu, S.; McLaren, W. M.; Matubayasi, N. The Hofmeister Series and Protein-Salt Interactions. J. Chem. Phys. 2006, 124, 234905.
- (S10) Weerasinghe, S.; Smith, P. E. A Kirkwood-Buff derived force field for sodium chloride in water. J. Chem. Phys. 2003, 119, 11342–11349.
- (S11) Kalcher, I.; Dzubiella, J. Structure-thermodynamics relation of electrolyte solutions. *J. Chem. Phys.* **2009**, *130*, 134507.
- (S12) Klasczyk, B.; Knecht, V. Kirkwood-Buff derived force field for alkali chlorides in simple point charge water. J. Chem. Phys. 2010, 132, 024109.
- (S13) Fyta, M.; Netz, R. R. Ionic force field optimization based on single-ion and ion-pair solvation properties: Going beyond standard mixing rules. *J. Chem. Phys.* **2012**, *136*, 124103.
- (S14) Newman, K. E. A Kirkwood-Buff theoretical approach to Debye-Hückel theory. J. Chem. Soc., Faraday Trans. 1 1989, 85, 485–492.
- (S15) Debye, P.; McAulay, J. Das Elektrische Feld Der Ionen Und Die Neutralsalzwirking (The electric field of ions and the action of neutral salts). *Physik. Z.* **1925**, *26*, 22–29.
- (S16) Born, M. Volumen und hydratationswärme der Ionen (Volume and hydration heat of ions). Z. Physik 1920, 1, 45–48.