

Amine-Promoted *anti*-Markovnikov Addition of 1,3-Dicarbonyl Compounds with Terminal Alkynes under Rhenium Catalysis

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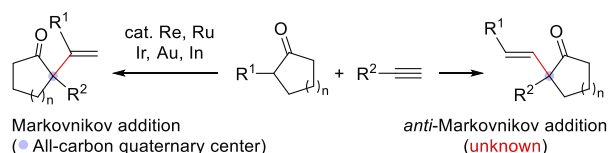
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KEYWORDS: Rhenium, *anti*-Markovnikov Addition, 1,3-Dicarbonyl Compound, Vinylidene, Amine

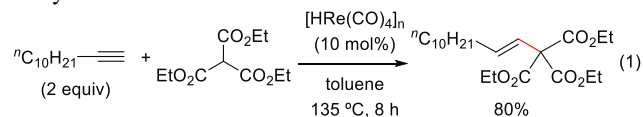
ABSTRACT: Amines have been identified to greatly accelerate the intermolecular *anti*-Markovnikov addition of carbon nucleophiles to unactivated terminal alkynes. Using a combination of [ReBr(CO)₃(thf)]₂ and ^tPr₂NEt, construction of cyclic all-carbon quaternary centers was achieved with various 1,3-ketoesters, diketones, and diesters with lower catalyst loading under milder conditions. The type of addition could be easily controlled by choice of additive, highlighting the unique features of rhenium catalysis

Rhenium complexes are powerful tools for the construction of complicated molecules from simple starting materials. Especially, the unique reactivity of rhenium carbonyl complexes has enabled unique carbon–carbon bond formation based on the activation of both nonpolar alkynes and polar carbonyl compounds.¹ Because group 7 rhenium complexes are located in the middle of early and late transition metal complexes in the periodic table, we rationalized this versatile reactivity was resulted from the hard and soft Lewis acidity that rhenium possesses.^{1a,j} Thus, tight control of the unique Lewis acidic nature of rhenium complexes was hypothesized to be possible through the proper choice of base additives. The present study demonstrates the ability to control regiodivergent addition reactions of carbon nucleophiles to unactivated terminal alkynes for the construction of cyclic all-carbon quaternary centers.² This type of structure is found in a wide range of natural products, biologically active compounds, and pharmaceuticals. Although several synthetic approaches for creating these fundamental structures have been reported, most required reactive halogenated or metalated reagents, and produced stoichiometric amounts of inorganic salt waste.² Catalytic intermolecular addition of enol equivalents with terminal alkynes can circumvent this disadvantage, and produce useful carbonyl and alkenyl functionalities in the resulting adducts.³ From a mechanistic perspective, the addition pattern of terminal alkynes can be classified into two types; Markovnikov and *anti*-Markovnikov addition (Scheme 1).⁴ Although rhenium, ruthenium, iridium, gold, and indium catalysts effectively promote the addition reaction, they provide only Markovnikov adducts.⁵ Thus, the intermolecular *anti*-Markovnikov addition of carbon nucleophiles remains a challenge.

Scheme 1. Construction of an All-Carbon Quaternary Center by Addition of Enol Equivalents with Terminal Alkynes



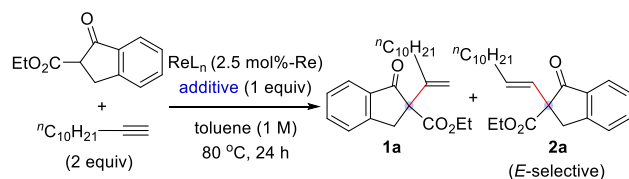
Recently, we reported the rhenium-catalyzed *anti*-Markovnikov addition of carbon nucleophiles with unactivated terminal alkynes.^{1g} In contrast to the well-investigated *anti*-Markovnikov addition of heteroatom nucleophiles, this was a rare example of the corresponding *intermolecular* addition of carbon nucleophiles.⁶ While this study provided a significant advance for regioselective carbon–carbon bond formation, limited substrate scope (applicable only methanetricarboxylates and symmetrically substituted 1,3-diester), high catalyst loading (10 mol% of [HRe(CO)₃]_n required), and elevated temperatures (135–150 °C) remained obstacles for the practical use of the reaction (Eq 1). The *anti*-Markovnikov addition of various 1,3-dicarbonyl compounds, including 1,3-ketoesters and diketones, using a combination of [ReBr(CO)₃(thf)]₂ and ^tPr₂NEt under relatively mild conditions is described in this study.



Our study commenced with an attempt to perform *anti*-Markovnikov addition of indanone ester with 1-dodecyne at 80 °C with a reduced amount of rhenium catalyst (2.5 mol%). The [HRe(CO)₃]_n, which was the best catalyst in our previous study,^{1g} afforded a mixture of Markovnikov adduct **1a** and *anti*-Markovnikov adduct **2a** in 27% and 31% yield, respectively (Table 1, entry 1). In contrast, Markovnikov adduct **1a** was obtained selectively in 94% yield in the presence of [ReBr(CO)₃(thf)]₂ catalyst (entry 2). Previous our study on the *anti*-Markovnikov addition of methanetricarboxylates revealed that a rhenium vinylidene species, generated *via* 1,2-hydrogen shift of a terminal alkyne, was a key intermediate for controlling the regio- and stereoselectivity of the overall addition reaction, and its generation is included as a rate determining step.^{1g} Thus, the ability of additives to accelerate the rate of the 1,2-hydrogen shift was next investigated.⁷ The results revealed that the use of Et₃N completely switched the regioselectivity to provide **2a** in 68% yield without formation of **1a** (entry 3). The yield of **2a** was highest in *n*-octane, although

reaction in other solvents, such as chlorobenzene, cyclohexane, and 1,4-dioxane, also afforded **2a** selectively (entry 4, and Table S1 in SI for details of reaction conditions optimization). After surveying various additives, the combination of bulky *tert*- and *sec*-alkylamines with $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ were most effective to promote selective *anti*-Markovnikov addition (entries 4–6). Among them, Pr_2NEt gave the best result to afford **2a** in 80% yield as a single regio- and stereoisomer (entry 6). Decreasing the concentration from 1.0 to 0.25 M to prevent the competitive undesirable oligo- or polymerization of 1-dodecyne slightly increased the yield of **2a** (entry 7). Finally, an isolated yield of 94% was achieved when 0.5 equiv of 1-dodecyne was added at the beginning and additional 1-dodecyne was added three times (0.5 equiv each) at 3 h intervals (entry 8).⁸ Markovnikov adduct **1a** was obtained selectively in 94% yield under the same conditions without Pr_2NEt , which demonstrates that addition regioselectivity could be easily and completely controlled by the choice of additive (entry 9).

Table 1. Effect of Amines on Regioselective Addition of Indanone Ester with 1-Dodecyne



entry	ReL_n	additive	Yield of 1a ^a	Yield of 2a ^a
1	$[\text{HRe}(\text{CO})_3]_n$	—	27%	31%
2	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	—	94%	0%
3	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	Et_3N	0%	68%
4 ^b	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	Et_3N	0%	78%
5 ^b	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	DABCO	0%	1%
6 ^b	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	Pr_2NEt	0%	80%
7 ^c	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	Pr_2NEt	0%	86%
8 ^{c,d}	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	Pr_2NEt	0%	96% (94%)
9 ^{c,d}	$[\text{ReBr}(\text{CO})_3(\text{thf})]_2$	—	96% (94%)	0%

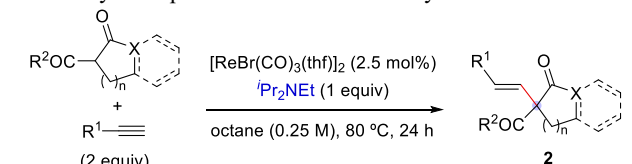
^aDetermined by ^1H NMR. The value in parentheses is the isolated yield.

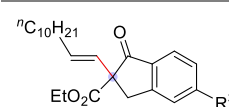
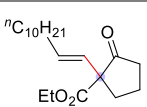
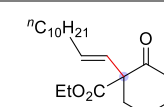
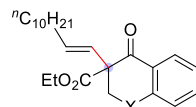
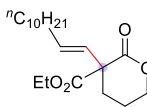
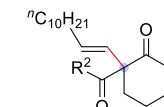
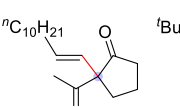
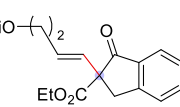
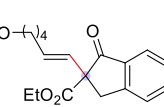
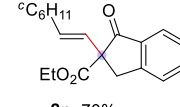
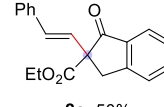
^bIn octane (1 M). ^cIn octane (0.25 M). ^d0.5 equiv of alkyne was added at the beginning and additional alkyne was added three times (0.5 equiv each) at intervals of 3 h.

Under the reaction conditions listed in Table 1, entry 8, indanone ester having an electron-withdrawing bromo group did not reduce the efficiency of *anti*-Markovnikov addition to afford **2b**, whereas the yield of **2c** decreased with electron-donating methoxy group substituted indanone ester (Table 2). Selective addition at the most acidic C–H bond was observed in the reaction of cyclopentanone ester to provide the corresponding *anti*-Markovnikov adduct **2d** in moderate yield. Addition of six membered-ring 1,3-ketoesters, such as cyclohexanone, α -tetralone, and 4-chromanone esters, also proceeded with complete regio- and stereoselectivity to furnish **2e**, **2f**, and **2g**, respectively. While reaction retarded with cyclic 1,3-diesters leading to **2h** even at 120 °C, addition reaction with 1,3-diketones proceeded smoothly to form the corresponding adducts **2i**, **2j**, and **2k** in good yields. In all cases, adducts were obtained as single regio- and stereoisomers. Functional groups, such as silyloxy and acetoxy groups, in terminal alkynes were tolerated to provide **2l** and **2m** in high yields. Sterically congested cyclohexylacetylene and phenylacetylene provided

the corresponding adducts **2n** and **2o**, respectively. Note that arylacetylenes could not be used as terminal alkynes in the previous $[\text{HRe}(\text{CO})_3]_n$ -catalyzed system due to their rapid oligomerization.^{1g}

Table 2. Rhenium-Catalyzed *anti*-Markovnikov Addition of 1,3-Dicarbonyl Compounds with Terminal Alkynes^a



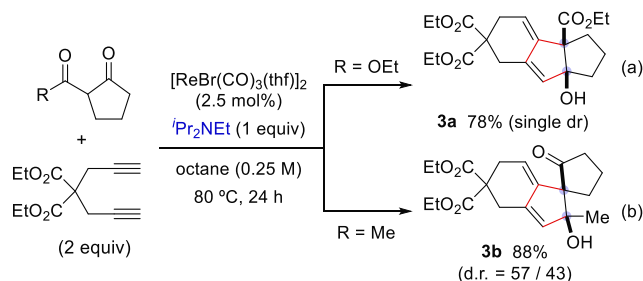
 2a ($\text{R}^3 = \text{H}$) 94% 2b ($\text{R}^3 = \text{Br}$) 89% 2c ($\text{R}^3 = \text{OMe}$) 63%	 2d 56% ^b	 2e 65% ^b
 2f ($\text{X} = \text{CH}_2$) 58% ^b 2g ($\text{X} = \text{O}$) 64% ^b	 2h 40% ^c	 2i ($\text{R}^2 = \text{Me}$) 80% ^d 2j ($\text{R}^2 = \text{Ph}$) 60% ^d
 2k 70% ^d	 2l 94%	 2m 80%
 2n 70%	 2o 50%	

^a0.5 equiv of alkynes were added at the beginning and additional alkynes were added three times (0.5 equiv each) at intervals of 2 h. ^b Pr_2NEt (5 equiv). ^cIn dioxane at 120 °C. 15% of starting 1,3-diesters was recovered. ^dIn octane (1 M).

Combination of $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ catalyst and Pr_2NEt also promoted the sequential cyclization of 1,6-diynes initiated by *anti*-Markovnikov addition.^{1j} The reaction enabled construction of vicinal quaternary tetrasubstituted carbon centers.⁹ Reaction with cyclopentanone ester gave tricycle **3a** as a single diastereomer in 78% yield *via* the sequential attack of the resulting alkenylrhenium species to the alkynyl and ethoxycarbonyl groups (Scheme 2(a), see Scheme S1 in SI for the mechanism).¹⁰ Use of a 1,3-diketone, such as 2-acetylcyclopentanone ester, as a carbon nucleophile gave spirocycle **3b** in 88% yield as a mixture of two diastereomers (Scheme 2(b)). In this case, the second nucleophilic attack of the alkenylrhenium species occurred selectively at the *exo*-ketocarbonyl group. These unique cyclization modes cannot be achieved without rhenium catalysts. Group 7 rhenium complexes may possess both soft and hard Lewis acidity, because they are situated in the middle of early and late transition metals in the periodic table. This characteristic might be key to the sequential interaction and attack of both nonpolar alkynyl groups and polar carbonyl groups.

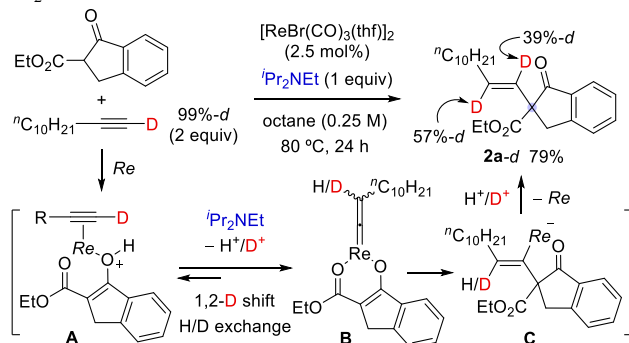
A control experiment using deuterium-labeled 1-dodecyne gave a mixture of deuterated *anti*-Markovnikov adducts **2a–d** (Scheme 3). This result indicates that rapid and competitive H/D-scrambling with the active methylene proton of

Scheme 2. Construction of Adjacent All-Carbon Quaternary Centers *via* Sequential Cyclization Initiated by *anti*-Markovnikov Addition to 1,6-Diynes



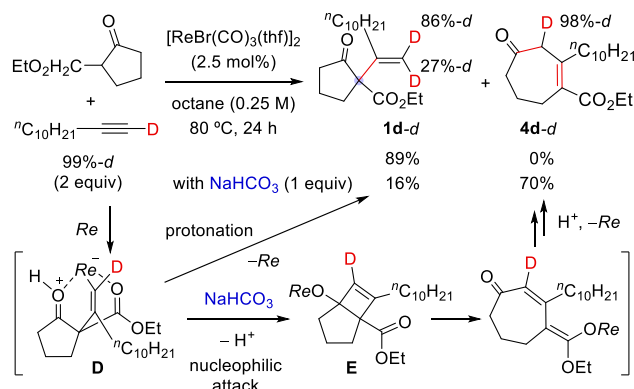
cyclopentanone ester occurred in the course of 1,2-D shifts, leading to the rhenium vinylidene intermediate **B** from π -alkyne complex **A**.^{11–13} Equilibrium developed between **A** and **B**, with formation of **B** preferred in the presence of amine (*vide infra*). The partial deuterium atom incorporation at the β -position of a decyl group of **2a-d** can be explained by quenching of the alkenylrhenium intermediate **C** by the deuterated cyclopentanone ester.

Scheme 3. Deuterium Labeling Experiments in the Presence of $t\text{Pr}_2\text{NEt}$



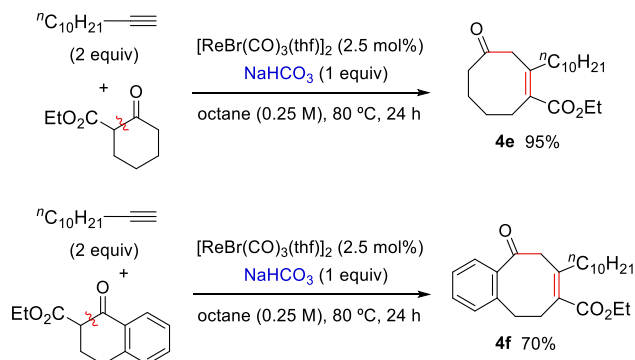
Experimental results in Table 1 demonstrated that the regioselectivity of the addition reaction could be adjusted by the proper choice of additive. A similar reactivity switch was observed for addition reactions of other 1,3-dicarbonyl compounds. In addition, we found selective formation of ring-expanded adduct **4** by using NaHCO_3 as an additive.¹⁴ A control experiment using cyclopentanone ester and deuterium-labeled 1-dodecyne in the presence and absence of NaHCO_3 provided ring-expanded adduct **4d-d** and Markovnikov adduct **1d-d**, respectively (Scheme 4).¹⁵ This result demonstrates that

Scheme 4. Deuterium Labeling Experiments in the Presence/Absence of NaHCO_3



formation of both **1d** and **4d** did not proceed *via* 1,2-hydrogen shift leading to a vinylidene intermediate. The incorporation of deuterium atom predominantly into the *anti*-position of the decyl group in **1d-d** indicates that Markovnikov addition proceeded mainly *via* intermediate **D**.¹⁶ Formation of ring-expanded adduct **4d-d** was attributed to intramolecular nucleophilic cyclization of alkenylrhenium intermediate **D**, followed by reductive elimination, ring-opening of intermediate **E** *via* *retro*-Aldol reaction, and isomerization. Addition of NaHCO_3 may prevent the rapid protonation of intermediate **D** leading to **1d-d**. This ring expanded addition in the presence of NaHCO_3 was confirmed to occur with other 1,3-ketoesters, such as cyclohexanone and tetralone esters, to afford 4-oxo-1-cyclooctene derivatives **4e** and **4f** in high yields (Scheme 5).

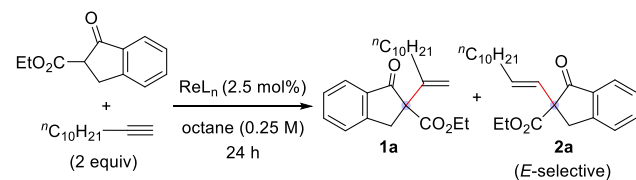
Scheme 5. NaHCO_3 -Promoted Ring-Expanded Addition of 1,3-Ketoesters with 1-Dodecyne



This unique switching of reactivity was observed only with $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ and $\text{ReBr}(\text{CO})_5$ catalysts. Other rhenium complexes, including $[\text{HRe}(\text{CO})_3]_n$ and $\text{Re}_2(\text{CO})_{10}$, which were effective for our previous *anti*-Markovnikov addition reactions of methanetricarboxylates,^{18j} gave mixtures of adducts and failed to switch reactivity. This difference in reactivity was thought to be due to the soft Lewis acidic nature of rhenium complexes. Because Markovnikov adduct **1a** *via* direct addition of indanone ester was obtained in high yield with $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ rather than $\text{Re}_2(\text{CO})_{10}$, the former may possess stronger Lewis π -acidity toward carbon–carbon triple bonds (Table 3, entry 1 *vs* 3). Thus, greater change in catalytic performance should be observed for $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ upon treatment of Lewis base ($t\text{Pr}_2\text{NEt}$). This characteristic is key for the current unique regioselectivity switching of addition reactions. Coordination of $t\text{Pr}_2\text{NEt}$ decreased the Lewis acidity of the rhenium center, and slowed the direct addition of 1,3-dicarbonyl compounds to alkynes leading to Markovnikov adduct **1** (Figure 1). During the slow nucleophile addition, an amine-assisted 1,2-hydrogen shift occurred to form a rhenium vinylidene intermediate, leading to the selective formation of *anti*-Markovnikov adducts **2** with $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ in the presence of amine. Note that coordination of an amine to $[\text{ReBr}(\text{CO})_3(\text{thf})]_2$ in the presence of 1,3-dicarbonyl compounds can be monitored by ^1H NMR, which clearly showed that $t\text{Pr}_2\text{NEt}$ acted not only as a base to promote 1,2-hydrogen shift, but also as a ligand to adjust the Lewis acidity of an active rhenium center.¹⁷ Temperature was another factor in determining the regioselectivity, and *anti*-Markovnikov addition occurred preferentially over Markovnikov addition at high temperatures (Table 3). This implies that the equilibrium be-

tween the rhenium vinylidene complex and rhenium π -alkyne complex can be controlled by both the choice of additive and the temperature.

Table 3. Effect of Catalysts and Temperature on the Regioselective Addition Reaction of Indanone Ester^a



entry	ReL _n	additive	Yield of 1a ^a	Yield of 2a ^a
1	[ReBr(CO) ₃ (thf)] ₂	80	96%	2%
2		135	84%	10%
3	Re ₂ (CO) ₁₀	80	0%	0%
4		100	6%	6%
5		110	6%	24%
6		135	9%	70%

^aDetermined by ¹H NMR.

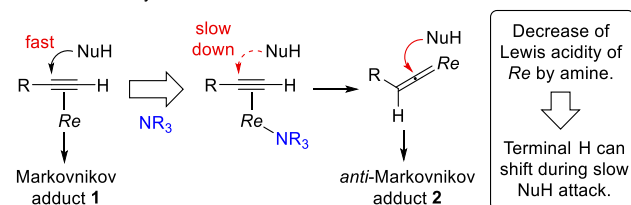


Figure 1. Rational for the Regiodivergent Addition Reaction

In summary, the current study demonstrated the ability to adjust the Lewis acid nature of rhenium complexes by selecting the appropriate base additive. The [ReBr(CO)₃(thf)]₂ catalyst was highly effective for the construction of complicated carbon frameworks with a cyclic all-carbon quaternary center *via* addition reaction of carbon nucleophiles to unactivated terminal alkynes. The proper choice of additives enabled control of the regio- and stereoselectivity of the addition reaction, and allowed the selective synthesis of Markovnikov adducts (no additive), *anti*-Markovnikov adducts (with ¹Pr₂NEt), and ring-expanded adducts (with NaHCO₃). The use of amines as additives resulted in application of various 1,3-dicarbonyl compounds, including 1,3-ketoesters, diketones, and diesters as substrates, which greatly expands the limited scope of *anti*-Markovnikov addition of carbon nucleophiles. Amine is thought to act as a base to promote 1,2-hydrogen shift as well as a ligand to adjust the Lewis acidity of [ReBr(CO)₃(thf)]₂, which accounts for the current unusual regiodivergent addition reaction. Further studies including the construction of all-carbon quaternary stereocenters are currently investigated in our laboratory.

ASSOCIATED CONTENT

Supporting Information. Experimental procedures, characterization data for all new compounds, and copies of ¹H and ¹³C NMR spectra. This material is available free of charge *via* the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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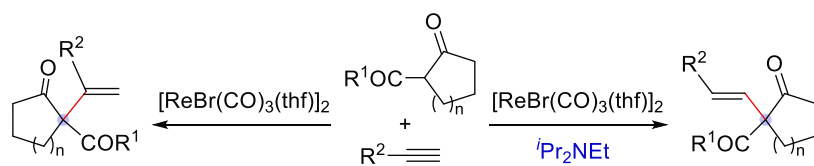
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- Fine-tuning of Lewis acid nature of rhenium complexes by base additives.
- Construction of cyclic all-carbon quaternary centers.